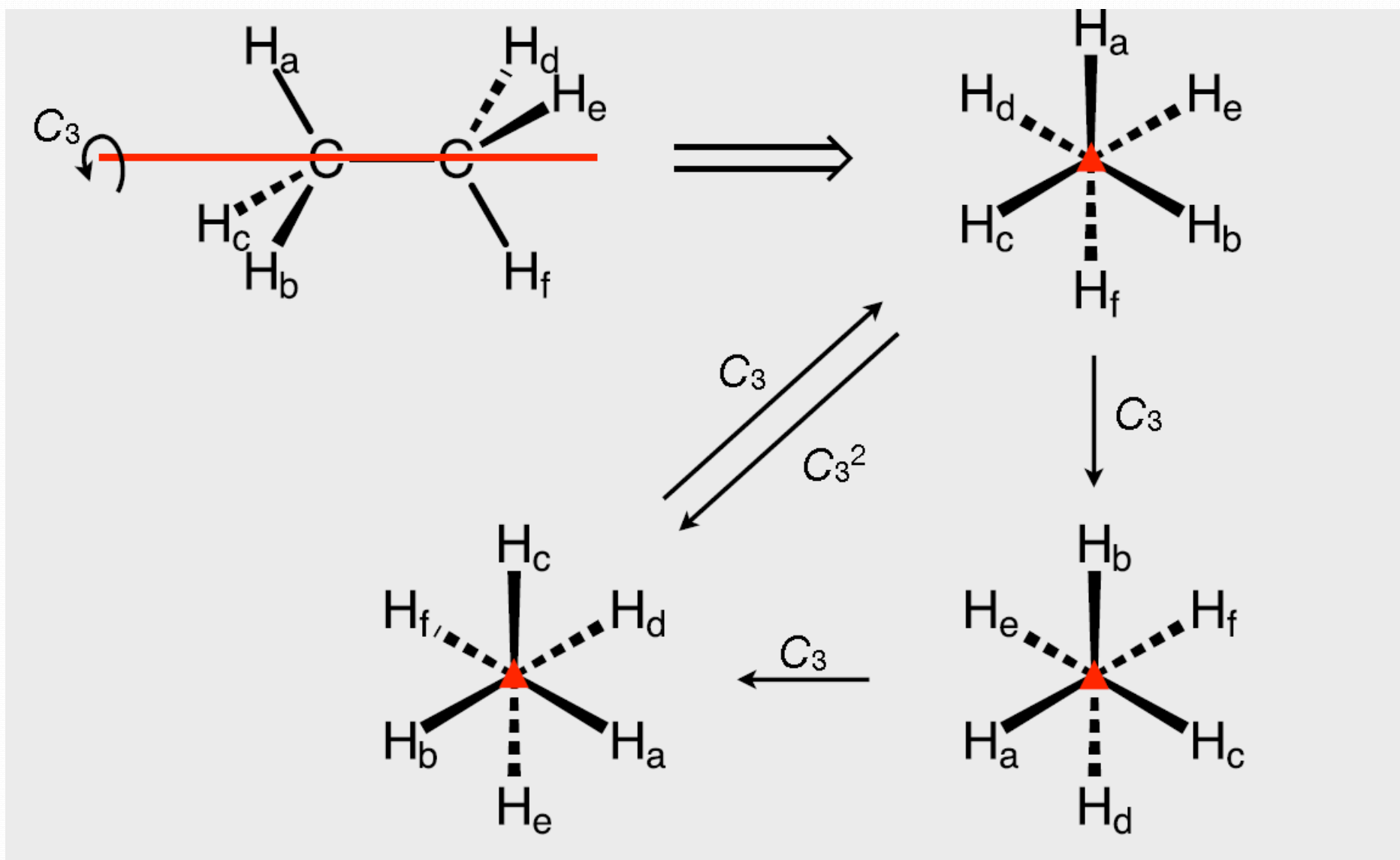


# **Symmetry and Point Groups**

Chapter 4

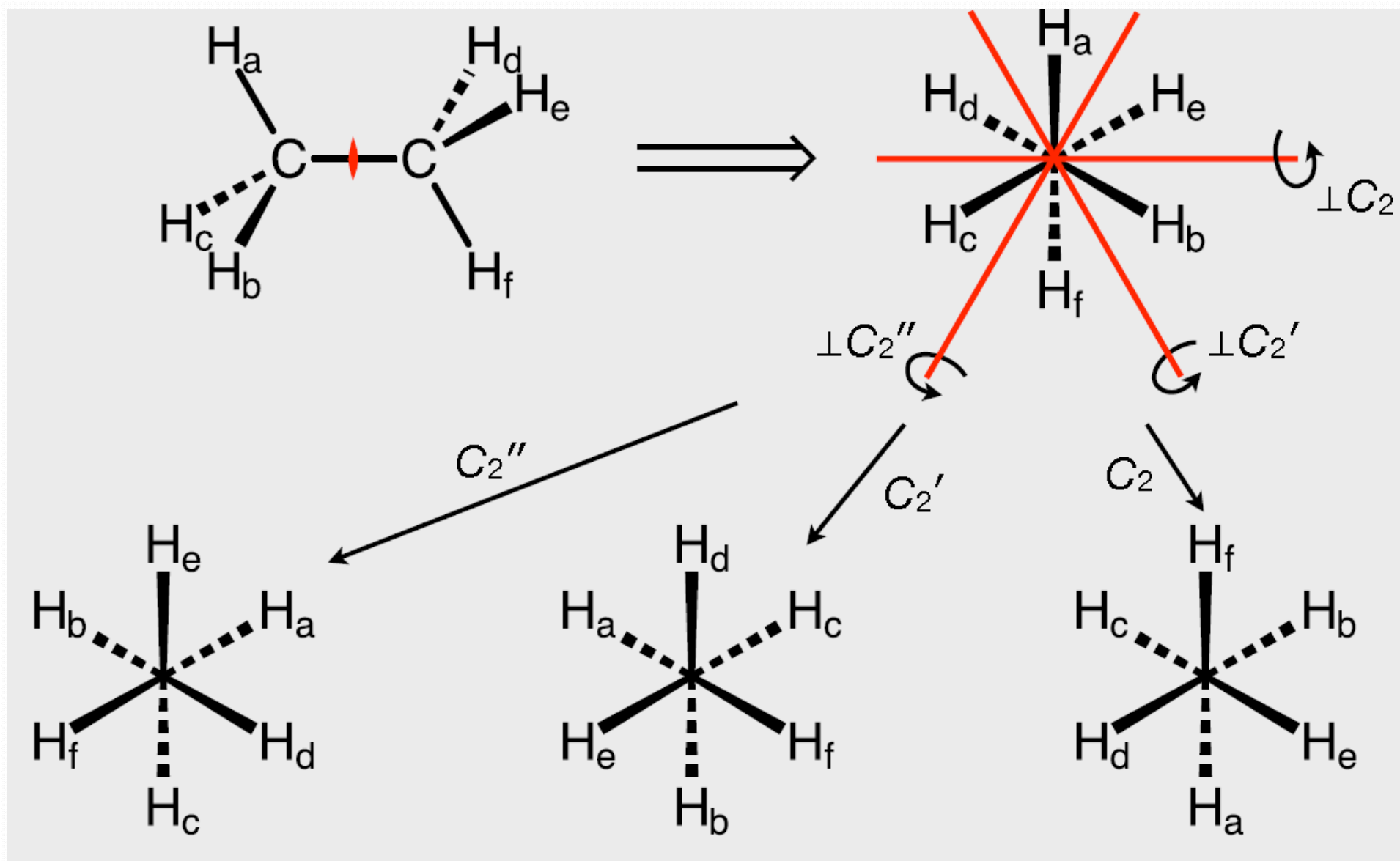
Monday, September 28, 2015

# Symmetry in Molecules: Staggered Ethane



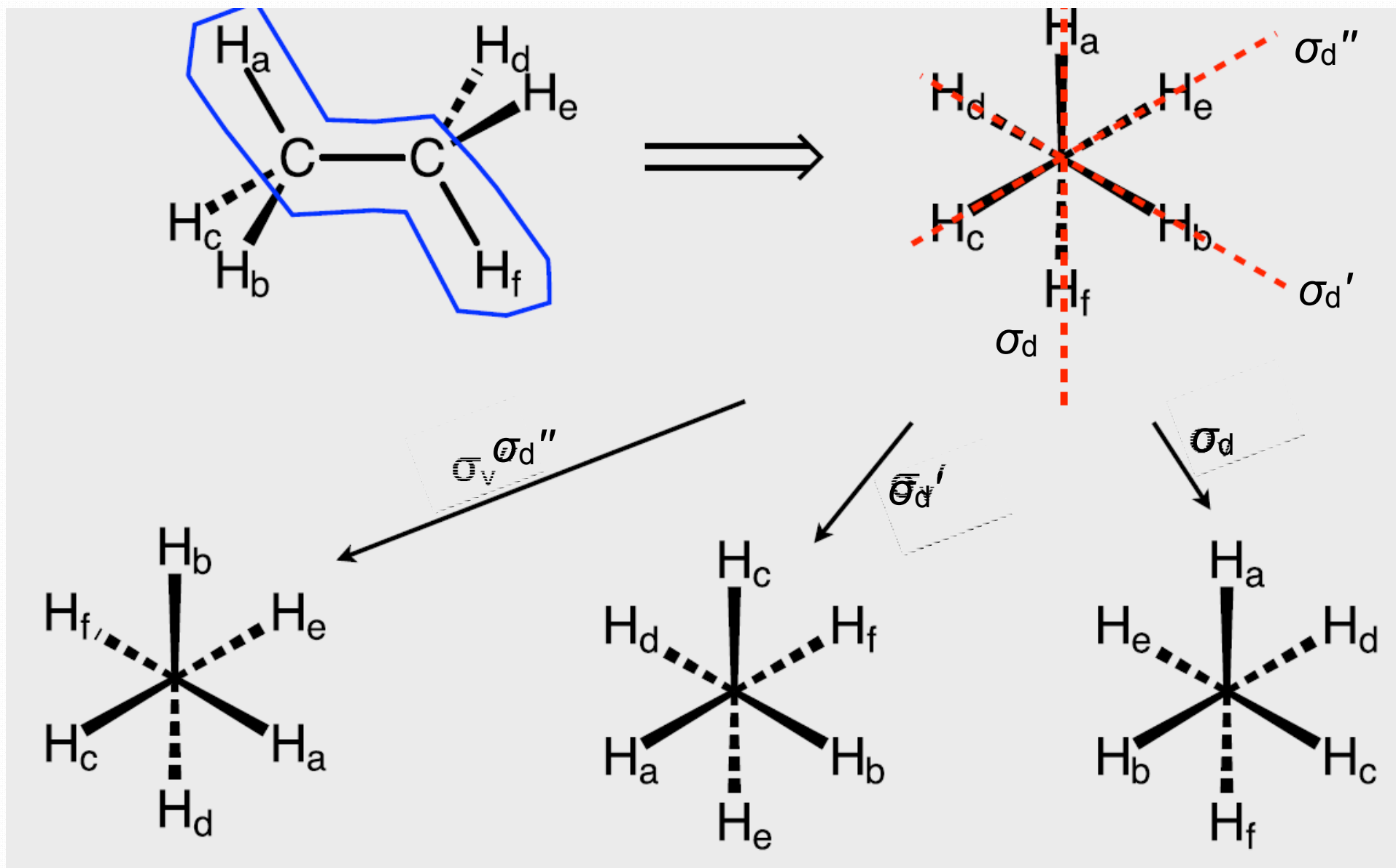
So far we can say staggered ethane has three operations:  $E$ ,  $C_3$ , and  $C_3^2$

# Symmetry in Molecules: Staggered Ethane



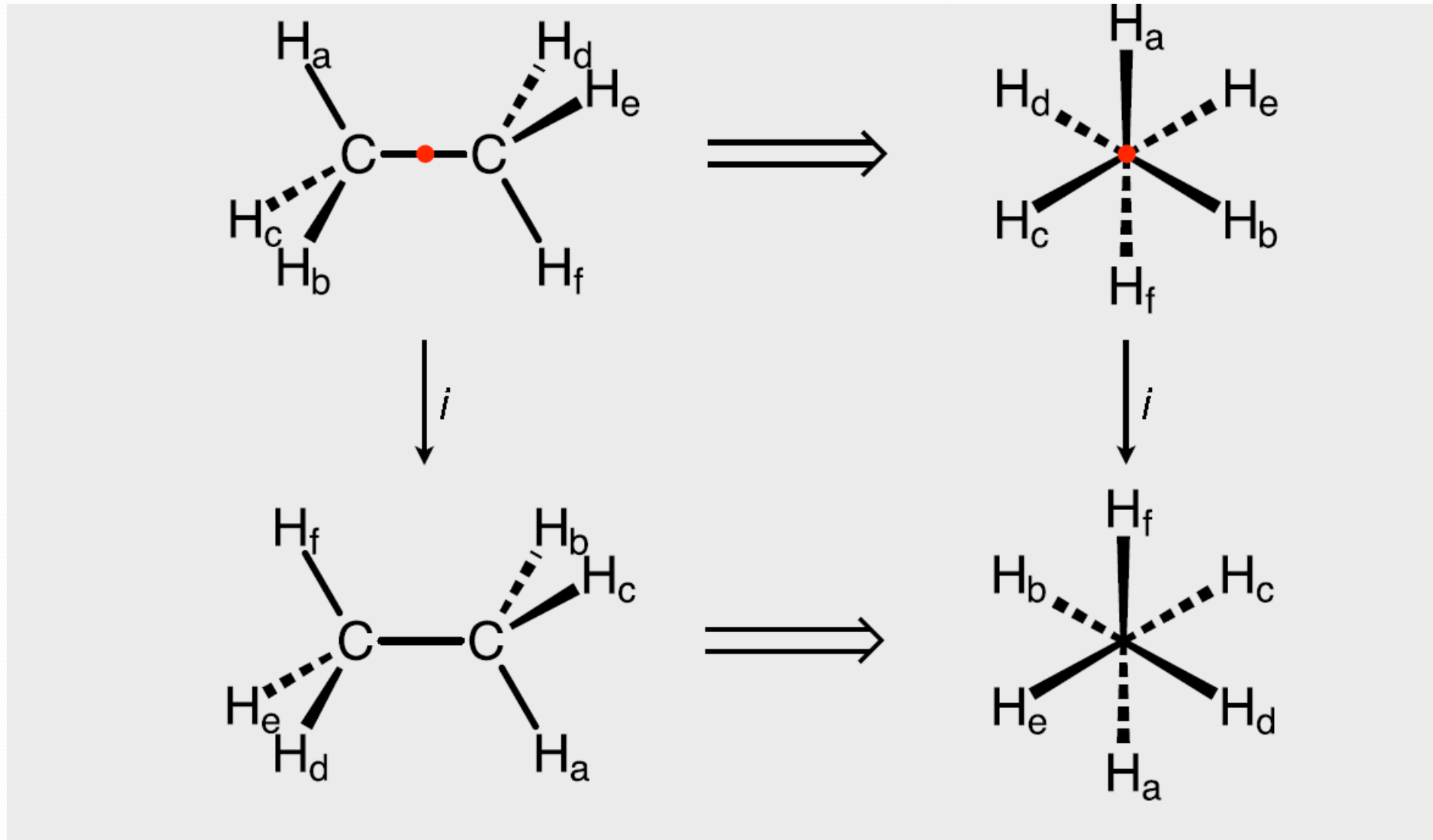
So we add three more operations:  $C_2$ ,  $C_2'$ , and  $C_2''$

# Symmetry in Molecules: Staggered Ethane



Now we've added three reflections:  $\sigma_d$ ,  $\sigma_{d'}$ , and  $\sigma_{d''}$   
Note that there is no  $\sigma_h$  for staggered ethane!

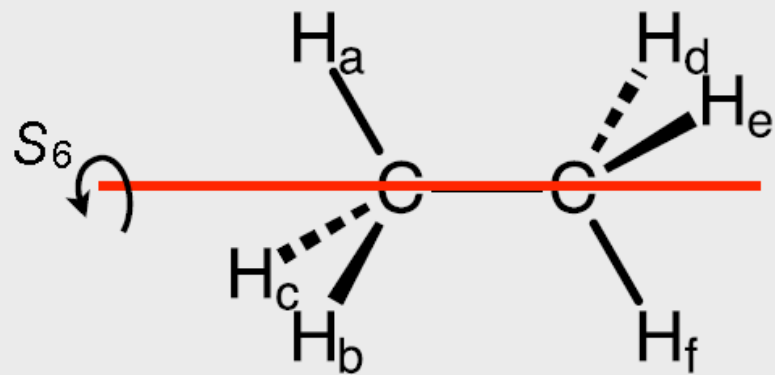
# Symmetry in Molecules: Staggered Ethane



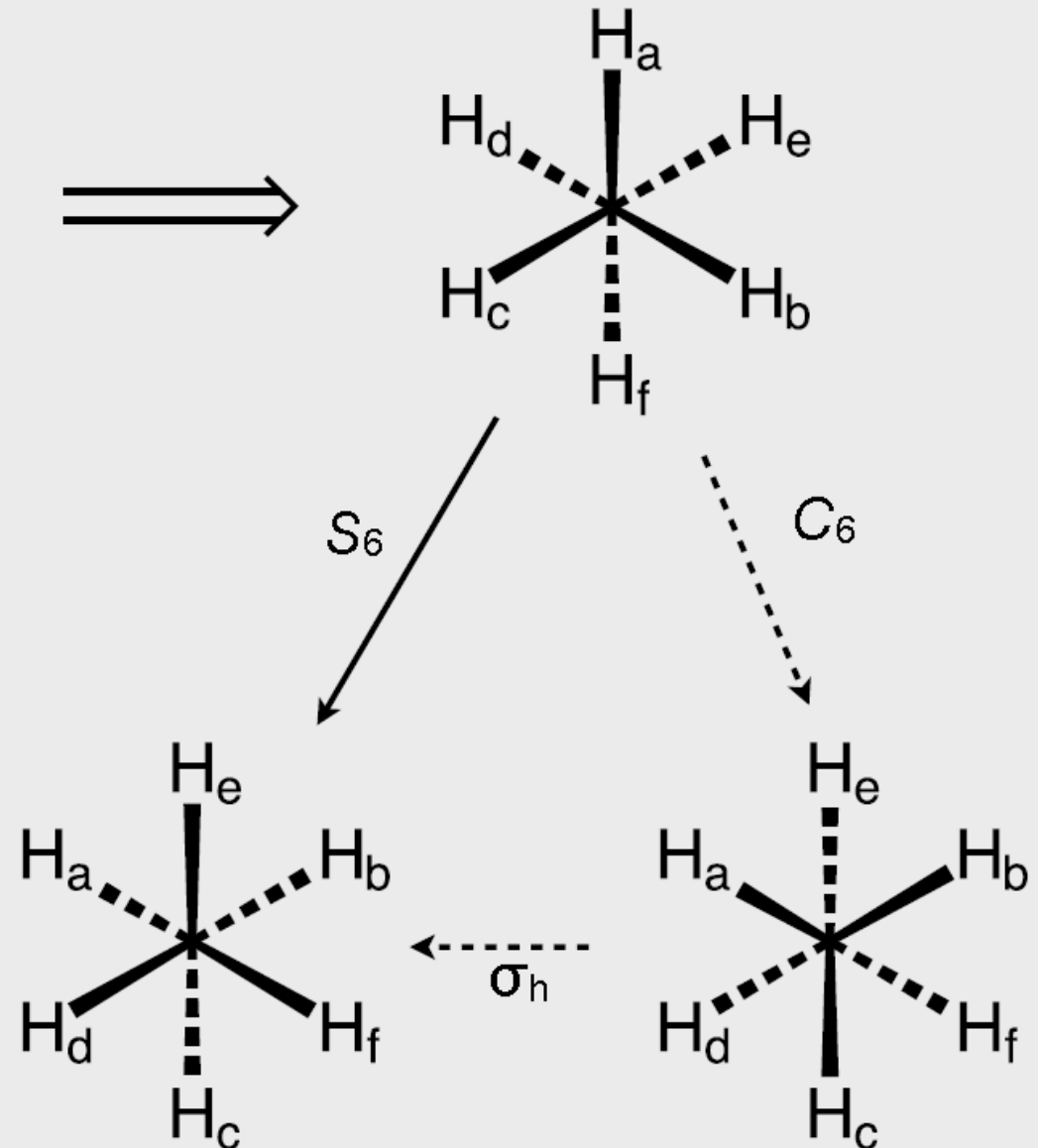
Ethane also has an inversion center that lies at the midpoint of the C-C bond (the center of the molecule).

# Symmetry in Molecules: Staggered Ethane

Finally, staggered ethane also has an improper rotation axis. It is an  $S_6$  ( $S_{2n}$ ) axis that is coincident with the  $C_3$  axis.

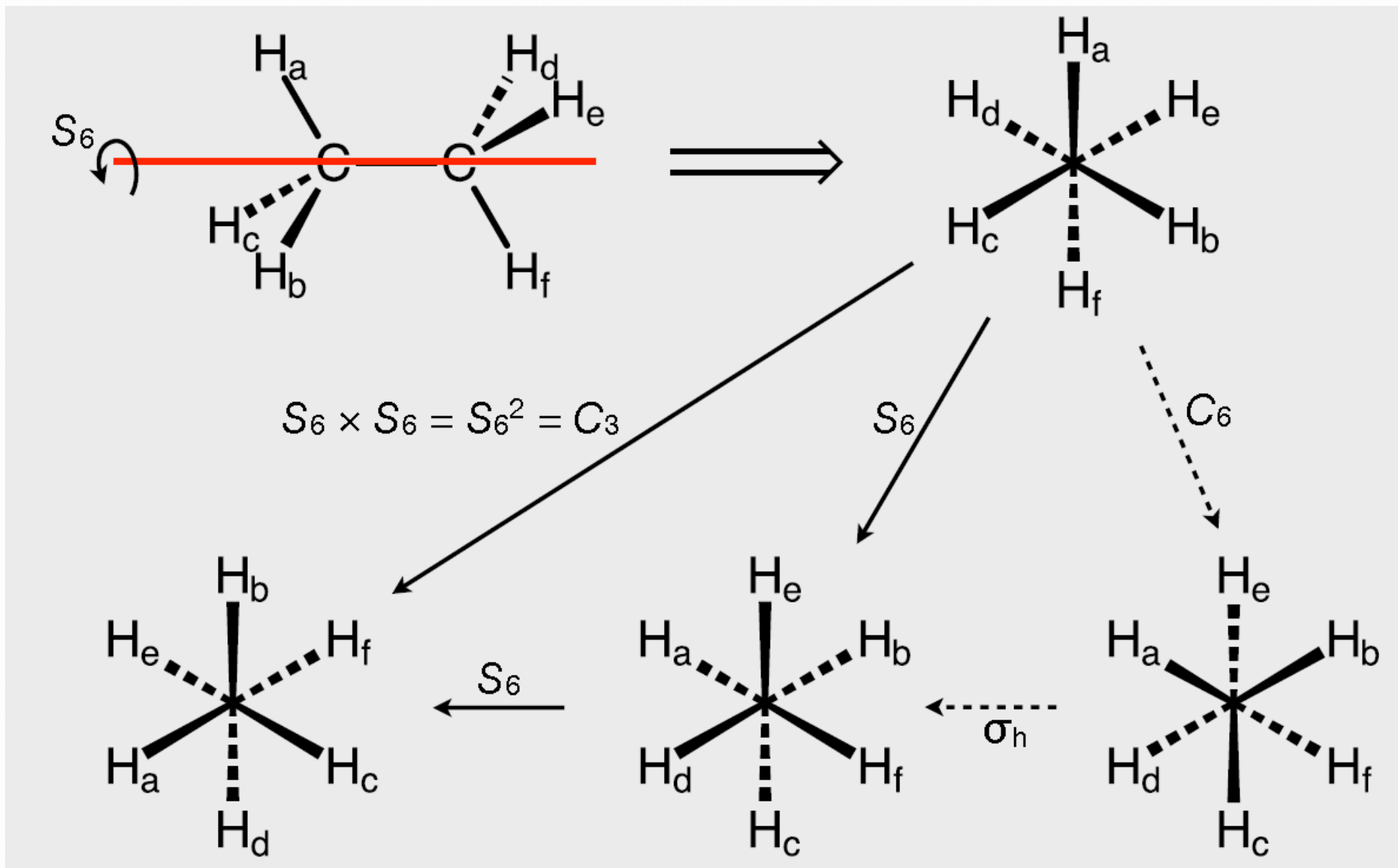


An  $S_6$  rotation is a combination of a  $C_6$  followed by a perpendicular reflection (i.e., a  $\sigma_h$ ).



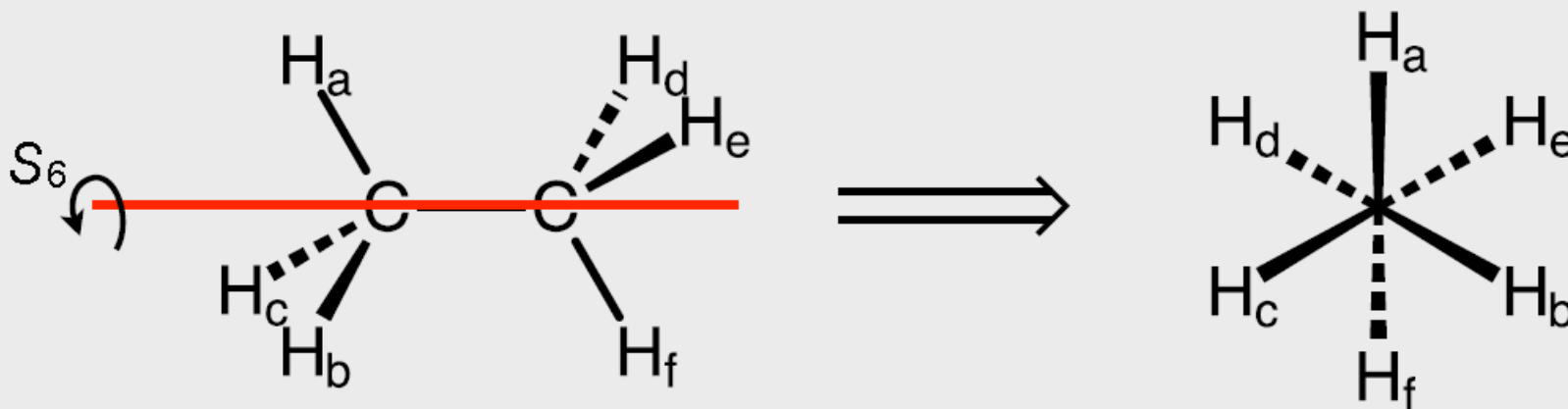
# Symmetry in Molecules: Staggered Ethane

Finally, staggered ethane also has an improper rotation axis. It is an  $S_6$  ( $S_{2n}$ ) axis that is coincident with the  $C_3$  axis.



# Symmetry in Molecules: Staggered Ethane

It turns out that there are several redundancies when counting up the unique improper rotations:



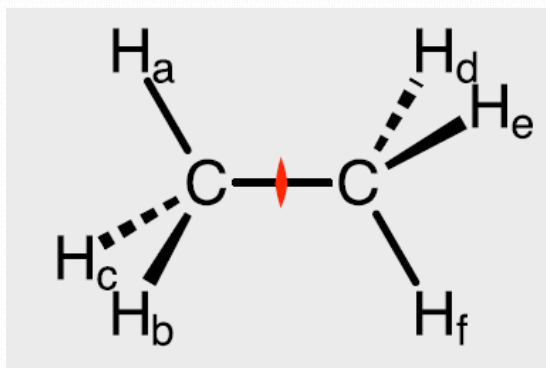
$S_6$ operation	equivalent operation
$S_6$	$S_6$
$S_6^2$	$C_3$
$S_6^3$	$i$
$S_6^4$	$C_3^2$
$S_6^5$	$S_6^5$
$S_6^6$	$E$

So the improper rotations add only two unique operations.



# Symmetry in Molecules: Staggered Ethane

Let's sum up the symmetry operations for staggered ethane:



Operation type	Number
Identity	1
Rotations	5 ( $2C_3 + 3C_2$ )
Reflections	3 ( $3\sigma_d$ )
Inversion	1
Improper Rotations	2 ( $S_6 + S_6^5$ )
<b>Total</b>	<b>12</b>

- These 12 symmetry operations describe completely and without redundancy the symmetry properties of the staggered ethane molecule.
- The complete set of symmetry operations possessed by an object defines its point group. For example, the point group of staggered ethane is  $D_{3d}$ .
- The total number of operations is called the order ( $h$ ) of a point group. The order is always an integer multiple of  $n$  of the principal axis. For staggered ethane,  $h = 4n$  ( $4 \times 3 = 12$ ).

# Summary

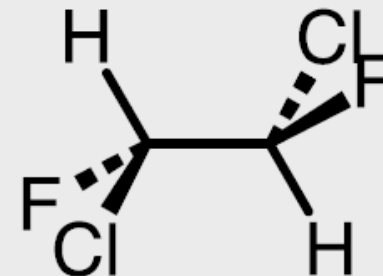
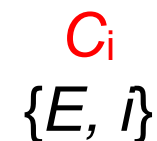
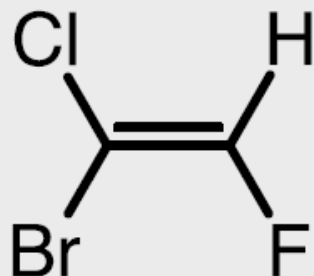
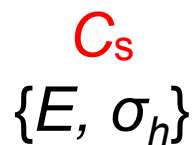
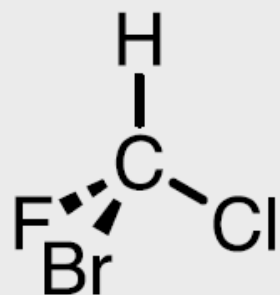
---

## Symmetry Elements and Operations

- **elements are imaginary points, lines, or planes within the object.**
- **operations are movements that take an object between equivalent configurations – indistinguishable from the original configuration, although not necessarily identical to it.**
- **for molecules we use “point” symmetry operations, which include rotations, reflections, inversion, improper rotations, and the identity. At least one point remains stationary in a point operation.**
- **some symmetry operations are redundant (e.g.,  $S_6^2 \equiv C_3$ ); in these cases, the convention is to list the simpler operation.**

# Low-Symmetry Point Groups

These point groups only contain one or two symmetry operations



# High-Symmetry Point Groups

These point groups are high-symmetry groups derived from Platonic solids

$T_d$

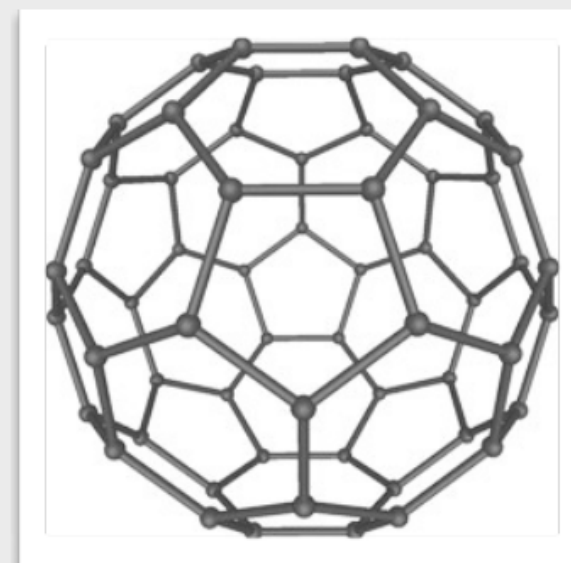
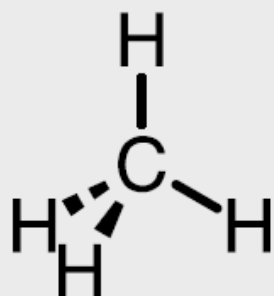
$$\{E, 8C_3, 3C_2, 6S_4, 6\sigma_d\} = 24$$

$O_h$

$$\{E, 8C_3, 6C_2, 6C_4, 3C_2, i, 6S_4, 8S_6, 3\sigma_h, 6\sigma_d\} = 48$$

$I_h$

$$\{E, 12C_5, 12C_5^2, 20C_3, 15C_2, i, 12S_{10}, 12S_{10}^3, 20S_6, 15\sigma\} = 120$$



**Buckminsterfullerene**  
**( $C_{60}$ )**

The five regular Platonic solids are the tetrahedron ( $T_d$ ), octahedron ( $O_h$ ), cube ( $O_h$ ), dodecahedron ( $I_h$ ), and icosahedron ( $I_h$ )

# High-Symmetry Point Groups

In addition to  $T_d$ ,  $O_h$ , and  $I_h$ , there are corresponding point groups that lack the mirror planes ( $T$ ,  $O$ , and  $I$ ).

Adding an inversion center to the  $T$  point group gives the  $T_h$  point group.

**TABLE 4.5** Symmetry Operations for High-Symmetry Point Groups and Their Rotational Subgroups

Point Group	Symmetry Operations									
$I_h$	$E$	$12C_5$	$12C_5^2$	$20C_3$	$15C_2$	$i$	$12S_{10}$	$12S_{10}^3$	$20S_6$	$15\sigma$
$I$	$E$	$12C_5$	$12C_5^2$	$20C_3$	$15C_2$					
$O_h$	$E$	$8C_3$	$6C_2$	$6C_4$	$3C_2(\equiv C_4^2)$	$i$	$6S_4$	$8S_6$	$3\sigma_h$	$6\sigma_d$
$O$	$E$	$8C_3$	$6C_2$	$6C_4$	$3C_2(\equiv C_4^2)$					
$T_d$	$E$	$8C_3$	$3C_2$				$6S_4$			$6\sigma_d$
$T$	$E$	$4C_3$	$4C_3^2$	$3C_2$						
$T_h$	$E$	$4C_3$	$4C_3^2$	$3C_2$		$i$	$4S_6$	$4S_6^5$	$3\sigma_h$	

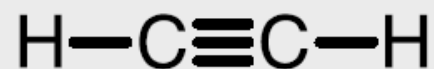
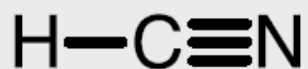
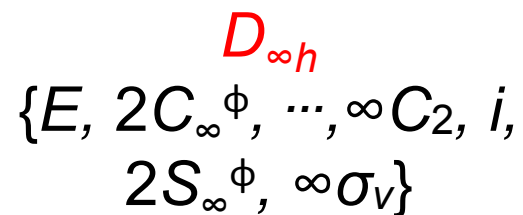
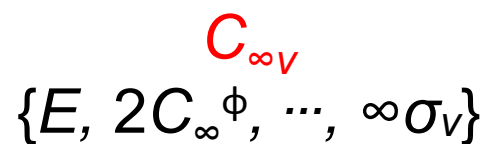
$T_h$  example:



# Linear Point Groups

---

These point groups have a  $C_\infty$  axis as the principal rotation axis



# D Point Groups

These point groups have  $nC_2$  axes perpendicular to a principal axis ( $C_n$ )

$D_n$

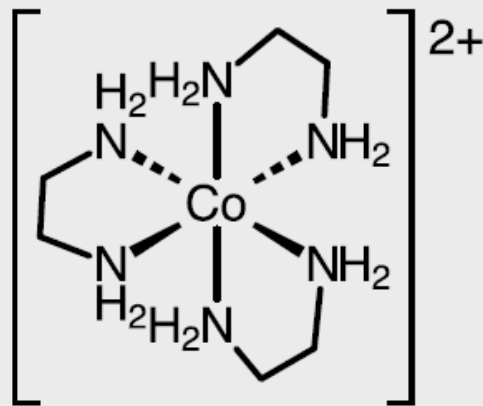
$\{E, (n-1)C_n, n \perp C_2\}$

$D_{nh}$

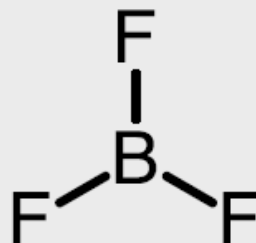
$\{\text{depends on } n, \text{ with } h = 4n\}$

$D_{nd}$

$\{\text{depends on } n, \text{ with } h = 4n\}$

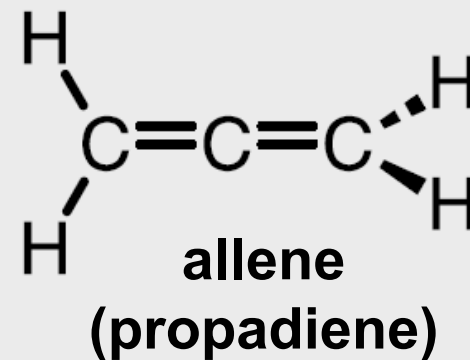


$D_3$



$D_{3h}$

$\{E, 2C_3, 3C_2, \sigma_h, 2S_3, 3\sigma_v\}$



$D_{2d}$

$\{E, 2S_4, C_2, 2C_2', 2\sigma_d\}$

# C Point Groups

These point groups have a principal axis ( $C_n$ ) but no  $\perp C_2$  axes

$C_n$

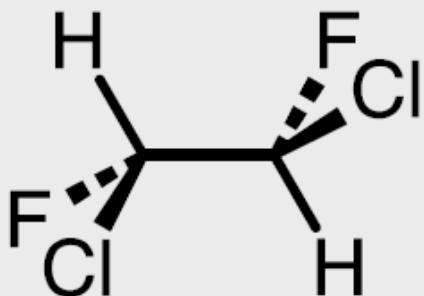
$\{E, (n-1)C_n\}$

$C_{nv}$

$\{E, (n-1)C_n, n\sigma_v\}$

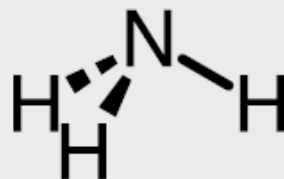
$C_{nh}$

$\{depends\ on\ n,\ with\ h = 2n\}$



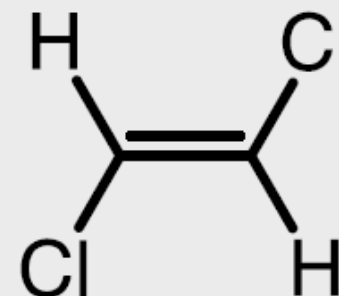
$C_2$

$\{E, C_2\}$



$C_{3v}$

$\{E, 2C_3, 3\sigma_v\}$



$C_{2h}$

$\{E, C_2, i, \sigma_h\}$



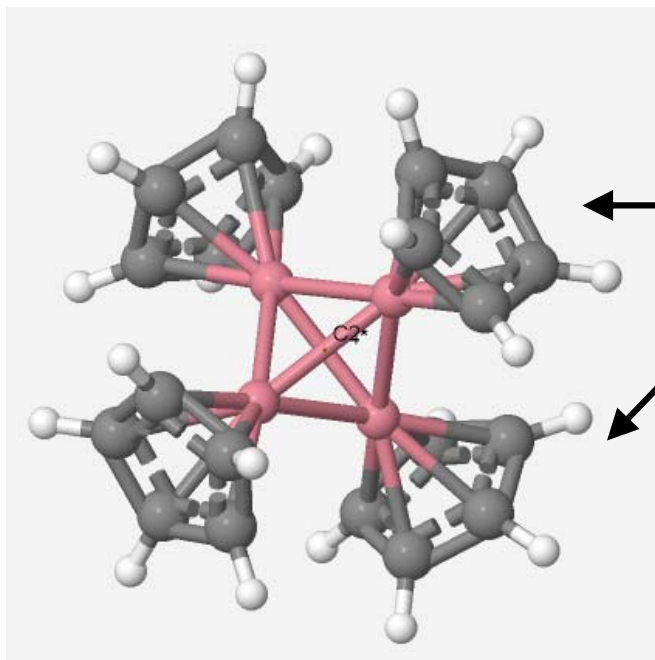
# S Point Groups

If an object has a principal axis ( $C_n$ ) and an  $S_{2n}$  axis but no  $\perp C_2$  axes and no mirror planes, it falls into an  $S_{2n}$  group

$S_{2n}$

{depends on  $n$ , with  $h = 2n$ }

$\text{Co}_4\text{Cp}_4$



← cyclopentadienyl (Cp)

ring =



$S_4$

{ $E, S_4, C_2, S_4^3$ }